

## Chemistry Stoichiometry Packet Answers

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Reactant  
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Unit 6: Reactions and Stoichiometry

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Chapter 12 Stoichiometry Packet Answer Key Answer: 4.93 x 10<sup>-5</sup> L or 49.3 μL In Example 12.2.1 and Example 12.2.2, the identity of the limiting reactant has been apparent: [Au(CN) 2]<sup>-</sup>, LaCl 3, ethanol, and para -nitrophenol.

Stoichiometry Packet Answers Chapter 12

Chemical Reactions and Stoichiometry. Given the equation 3A + B + C + D, you react 1 mole of A with 3 moles of B. True or false: B; is the limiting reactant because you have fewer moles of B than A. Answer: False. NaHCO 3; is the active ingredient in baking soda. How many grams of oxygen are in 0.35 g of. NaHCO 3? a) 0.012 g. b) 0.022 g. c) 0.067 g. d) 0.20 g

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Chapter 9 Stoichiometry

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