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space provided. 1. Identify whether the descriptions below describe an ideal gas or a real gas. a. Gas particles move in straight lines until they collide with other particles or the walls of their container.

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consist of large numbers of tiny particles that are far apart relative to their size 2.

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will . decrease b.

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1. 2. KMT - based on idea that particles of matter are always in motion. 2. 3. 1. Gases consist of large numbers of tiny particles that are far apart relative to their size.2.

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